

ESTD: 1967

P.B.No: 03



V.M.V.V.SANGHA'S

VIJAY MAHANTESH KRUPAPOSHIT

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National Level e-Quiz on "General Chemistry" Organized by Dept. Chemistry from 22-07-2020 to 25-08-2020.

Estd: 1967
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No. VMCH/ Chemistry/2019-20
Date: 25.08.2020

To,
The Coordinator, IQAC
V.M.K.S.R.Vastrad Arts, Science
And V.S.Bellihal Commerce College,
Hungund.

Sub: Submission of Report of National Level E-Quiz organized by Department of Chemistry. (IQAC Initiative)


Respected Sir,

With reference to the cited Subject above, I am Mr. SHRINIDHI. R. A. Assistant Professor in Chemistry, submitting the report on the National E-Quiz organized by the department of Chemistry on the 22nd July to 25th August 2020. 1146 faculty members, research scholars and Students have taken part in the E-Quiz.

Thanking You

Hungund,
25.08.2020

Your's faithfully
SHRINIDHI. R. A.
[SHRINIDHI. R. A.]
Assistant Professor
Dept of Chemistry


PRINCIPAL
Vijaya Mahantesh Krupaposhit
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Report of the e-Quiz

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List of Participants



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Re-accredited "B+" by NAAC at 3rd Cycle


(Affiliated to Rani Channamma University, Belagavi)


CERTIFICATE OF APPRECIATION

This is to certify that {{Full Name}} has successfully completed **National Level e-Quiz On "General Chemistry-2020"** an IQAC Initiative & Organized by Department of Chemistry of V.M.K S.R.VASTRAD ARTS, SCIENCE & V.S.BELLIHAL COMMERCE COLLEGE, HUNGUND, INDIA-587118 on {{Date}} .

{{Certificate id}}


(Dr. S.R. Golagond)
IQAC Coordinator


(Mr. Shrinidhi. R. A.)
HoD of Chemistry


(Prof. Smt. S.K. Math)
Principal

Certificate Format Issued to the Participants

National e-Quiz on General Chemistry

Greetings to All

Department of Chemistry, Vijaya Mahantesh Krupeshohit S. R. Vastrad Arts, Science & V. S. Bellihal Commerce College Hungund, Bagalkot Dist, Karnataka-587118 has organised an National Level e-Quiz on General Chemistry for Faculty members, Research Scholars and Students.

- Participants should fill the registration form carefully.
- Participants scores more than 40% in the quiz then they will receive an printable e-certificate through registered email id.

Please Stay Home, Stay Safe and Keep Learning CHEMISTRY.

With Best Regards,

Mr. Shrinidhi R.A.
HOD & Assistant Professor
Department of Chemistry
* Required

1. Email address *

2. Full Name in Capital Letters (with salutation Prof/Dr/Mr/Mrs/Ms) *

3. Participant Type *

Mark only one oval.

- UG STUDENT
- PG STUDENT
- FACULTY
- RESEARCH SCHOLAR
- Others

Google Form

4. Name of the College/ Institution/ University *

5. Department/ Combination *

6. Place

7. District

8. State

9. Country

10. Mobile Number *

National e-Quiz on General Chemistry

11. 1. The purest form of Iron is ¹ 2 points

Mark only one oval.

- A. Wrought iron
- B. Steel
- C. Pig iron
- D. Nickel steel

12. 2. Hydrogen Bomb is based on the principle of ² 2 points

Mark only one oval.

- A. Nuclear fission
- B. Nuclear Fusion
- C. Natural radioactivity
- D. Artificial radioactivity

13. 3. Among the following, the maximum covalent character is shown by the compound ³ 2 points

Mark only one oval.

- A. $MgCl_2$
- B. $FeCl_2$
- C. $SnCl_2$
- D. $AlCl_3$

14. 4. Which of the following is a non metal that remains liquid at room temperature? ⁴ 2 points

Mark only one oval.

- A. Chlorine
- B. Phosphorus
- C. Bromine
- D. Helium

15. 5. Boron cannot form which one of the following anions? * 2 points
- Mark only one oval.
- A. BO_2^-
 - B. BF_6^-
 - C. BH_4^-
 - D. $\text{B}(\text{OH})_4^-$
16. 6. S.I. unit of equivalent conductance * 2 points
- Mark only one oval.
- A. Ohm/ Cm
 - B. Siemens M^2 / equivalent
 - C. Siemens/ equivalent
 - D. mho/ Cm.
17. 7. Which one of the following is not a mixture * 2 points
- Mark only one oval.
- A. Air
 - B. Mercury
 - C. Milk
 - D. Cement
18. 8. Among the given nutrients, milk is a poor source of * 2 points
- Mark only one oval.
- A. Calcium
 - B. Protein
 - C. Vitamin C
 - D. Carbohydrates

19. 9. When Formaldehyde & Potassium hydroxide are heated, we get ? 2 points

Mark only one oval.

- A. Acetylene
 B. Methyl alcohol
 C. Methane
 D. Ethyl formate.

20. 10. A 'Breath Test' used by traffic police to check drunken driving uses? 2 points

Mark only one oval.

- A. Turmeric on filter paper
 B. Silica gel coated with silver nitrate
 C. Potassium permanganate - Sulphuric acid
 D. Potassium dichromate - Sulphuric acid

21. 11. The enzyme that converts glucose to ethyl alcohol is ? 2 points

Mark only one oval.

- A. Maltose
 B. Zymase
 C. Diastase
 D. Invertase

22. 12. Cobalt - 60 is commonly used in radiation therapy because it emits ? 2 points

Mark only one oval.

- A. Alpha rays
 B. beta rays
 C. gamma rays
 D. X-rays

23. 13. The M. K. S. system was first introduced by * 2 points
- Mark only one oval.
- A. Archimedes
 - B. Einstein
 - C. Newton
 - D. Giorgi

24. 14. What is the oxidation number of sulphur in H_2SO_3 * 2 points
- Mark only one oval.
- A. 2
 - B. -2
 - C. 1
 - D. -1

25. 15. 'Misch metal' is widely used in manufacture of which of the following * 2 points
- Mark only one oval.
- A. Material of car brake
 - B. Smoke detectors
 - C. Cigarette lighters
 - D. Emergency lights

26. 16. Temporary hardness is due to the presence of * 2 points
- Mark only one oval.
- A. Calcium carbonate
 - B. Calcium bi-carbonate
 - C. Magnesium sulphate
 - D. Calcium sulphate

27. 17. What are the elements present in Urea? 2 points

Mark only one oval.

- A. C, H, O
 B. C, N, O
 C. C, N, H
 D. C, O, N, H

28. 18. Ethanol is converted into ethoxy ethane? 2 points

Mark only one oval.

- A. By heating excess of ethanol with Conc. H_2SO_4 at $140^\circ C$
 B. By treating with conc. H_2SO_4 at room temperature
 C. By treating with conc. H_2SO_4 at $373K$
 D. By heating ethanol with excess of Conc. H_2SO_4 at $443K$

29. 19. Which one among the following is the heaviest? 2 points

Mark only one oval.

- A. 1 mole of water molecule
 B. 1 mole of ammonium molecules
 C. 1 mole of sodium atoms
 D. 1 mole of hydrogen molecules

30. 20. Which is the gas used for making vanaspathi from vegetable oils? 2 points

Mark only one oval.

- A. Hydrogen
 B. Nitrogen
 C. Argon
 D. Helium